

Biosynthesis of Plant Sterols: Stereochemistry of Hydrogen Elimination at C-7 in α -Spinasterol

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Summary It is indicated that the 7β -hydrogen of lanosterol derived from the 2-proS hydrogen of mevalonic acid (MVA) is removed in the biosynthesis of α -spinasterol.

It has been shown previously¹ that the biosynthesis of cholesterol in a rat liver preparation involves a 5,7-diene intermediate and that during the formation of this intermediate the 7β -hydrogen of lanosterol is eliminated. The

($^3\text{H} : ^{14}\text{C}$ ratio 6:30, atomic ratio 3:23:5). This proves the absence of tritium atom at the C-7 position in the biosynthesized α -spinasterol isolated from *Camellia sinensis*. It also indicates that the 7β -hydrogen of lanosterol derived from the 2-proS hydrogen of MVA is removed in the formation of this phytosterol. Since the biosynthesized α -spinasterol derived from [$2R$ - 2 - ^3H]MVA has four tritium atoms, it is inferred that there is no tritium at $22R$. This supports

The $^3\text{H} : ^{14}\text{C}$ ratios of the products obtained after the administration of $3R$ -[2 - ^{14}C -($2R$)- 2 - $^3\text{H}_1$]MVA ($^3\text{H} : ^{14}\text{C}$ ratio 9:37) and $3R$ -[2 - ^{14}C -($2S$)- 2 - $^3\text{H}_1$]MVA ($^3\text{H} : ^{14}\text{C}$ ratio 9:74) to a *Camellia sinensis* plant

	$3R$ -[2 - ^{14}C -($2R$)- 2 - $^3\text{H}_1$]MVA $^3\text{H} : ^{14}\text{C}$ Ratio	$3R$ -[2 - ^{14}C -($2S$)- 2 - $^3\text{H}_1$]MVA $^3\text{H} : ^{14}\text{C}$ Ratio
α -Spinasterol	7.27	5.90
3β -Acetoxy- $5\alpha(H)$ -stigmast-7-ene		5.78
3β -Acetoxy- $7\alpha,8\alpha$ -dihydroxy- $5\alpha(H)$ -stigmastane		5.75
3β -Acetoxy- 8α -hydroxy- $5\alpha(H)$ -stigmastan-7-one		6.30

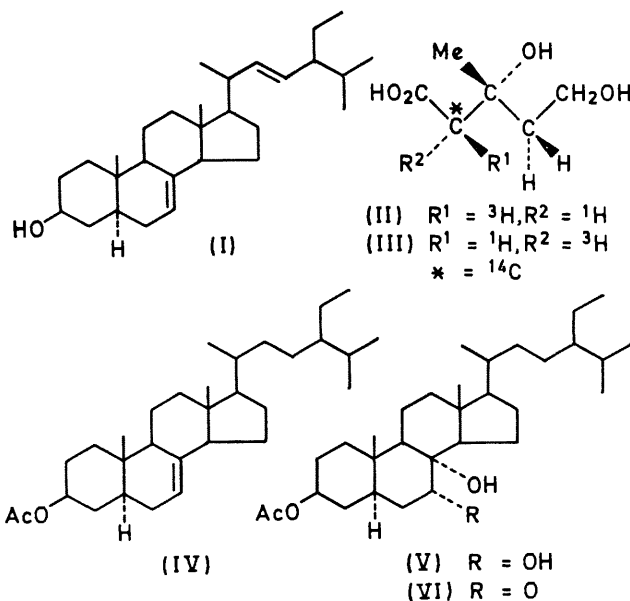
7β -hydrogen of lanosterol was shown also to be lost when the biosynthesis of poriferasterol was investigated in *Ochromonas malhamensis*.² However, in the yeast homogenate, the biosynthesis of C-27 cholesteryl analogues takes place by the elimination of the 7α -hydrogen via the formation of a Δ^7 -olefin.³

Since α -spinasterol (I) has a Δ^7 bond, we evaluated the mechanism of the formation of this double bond in this phytosterol.

$3R$ -[2 - ^{14}C -($2R$)- 2 - $^3\text{H}_1$]MVA (II) ($^3\text{H} : ^{14}\text{C}$ ratio 9:37, 50 μC of ^{14}C) or $3R$ -[2 - ^{14}C -($2S$)- 2 - $^3\text{H}_1$]MVA (III) ($^3\text{H} : ^{14}\text{C}$ ratio 9:74, 50 μC of ^{14}C) was administered to the *Camellia sinensis* plant.⁴ The plant was harvested after five weeks and α -spinasterol (I) was isolated from the non-saponifiable lipid.⁴ The carrier α -spinasterol was added and the sterol crystallized to constant specific activity and $^3\text{H} : ^{14}\text{C}$ ratio.

The $^3\text{H} : ^{14}\text{C}$ ratio obtained by the incorporation of [$2R$ - 2 - ^{14}C - 2 - ^3H]mevalonic acid was 7.27 (atomic ratio 3.82:5), indicating the presence of four tritium atoms. The $^3\text{H} : ^{14}\text{C}$ ratio obtained by the incorporation of [$2S$ - 2 - ^{14}C - 2 - ^3H]mevalonic acid was 5.90 (atomic ratio 3.03:5), indicating the presence of three tritium atoms.

The α -spinasterol biosynthesized from [$2S$ - 2 - ^{14}C - 2 - ^3H]MVA (III) was first acetylated and then hydrogenated over platinum oxide catalyst to give 3β -acetoxy- $5\alpha(H)$ -stigmast-7-ene⁵ (IV) ($^3\text{H} : ^{14}\text{C}$ 5.78, atomic ratio 2.96:5). The reaction of (IV) with osmium tetroxide in pyridine⁶ yielded 3β -acetoxy- $7\alpha,8\alpha$ -dihydroxy- $5\alpha(H)$ -stigmastane (V) ($^3\text{H} : ^{14}\text{C}$ ratio 5.75, atomic ratio 2.95:5). Oxidation of (V) (chromium trioxide-pyridine) gave 3β -acetoxy- 8α -hydroxy- $5\alpha(H)$ -stigmastan-7-one (VI) without the loss of tritium



the earlier observation² that the C-22 hydrogen originating from the 2-proR hydrogen of MVA is lost in the introduction of the Δ^{22} bond.

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¹ E. Caspi, J. B. Greig, P. J. Ramm, and K. R. Varma, *Tetrahedron Letters*, 1968, 3829.

² A. R. H. Smith, L. J. Goad, and T. W. Goodwin, *Chem. Comm.*, 1968, 926.

³ E. Caspi and P. J. Ramm, *Tetrahedron Letters*, 1969, 181.

⁴ R. K. Sharma, *Phytochemistry*, in the press.

⁵ D. H. R. Barton and J. D. Cox, *J. Chem. Soc.*, 1948, 1354.

⁶ L. F. Fieser and M. Fieser, "Steroids," Reinhold, New York, 1959, p. 240.